

Synthesis of leucine-enkephalin analogs containing α -amino squaric acid

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Abstract—Novel Leu-enkephalin analogs **10a–c** in which the Tyr¹, Gly², or Gly³ residue of Leu-enkephalin **9** was replaced with α -amino squaric acid analog Sq-Tyr **8b** or Sq-Gly **8a** were synthesized starting from **14** or **18**. Conformational analysis of **10a–c** together with its model compound **26** and their opioid receptor binding activity were evaluated.

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1. Introduction

Recently, a number of α -amino acid analogs (AAA) in which the carboxylic acid group of α -amino acid **1** is replaced with other acidic functional groups, for example, sulfinic acid **2**,¹ sulfonic acid **3**,² phosphinic acid **4**,³ phosphonic acid **5**,^{3,4,5} boronic acid **6**,⁶ and tetrazole **7**,⁷ have been developed (Fig. 1). Incorporation of these AAAs into biologically active peptides has received significant attention from bioorganic and medicinal chemists due to their important roles as enzyme inhibitors and haptens that generate catalytic antibodies.⁸ We have recently reported the synthesis of a new AAA **8** bearing a 4-hydroxy-2,3-dioxocyclobut-1-enyl (Sq) group (α -amino squaric acid, α -Asq) (Fig. 1).⁹ The Sq group is characterized by its planar structure, acidic OH group, and electron deficient property, and widely employed as a potent carboxylic acid surrogate in medicinal chemistry.^{10,11} We now report the synthesis of novel Sq-Gly- or Sq-Tyr-containing Leu-enkephalin (Leu-Enk) analogs **10a–c** bearing α -Asqs **8a,b** at the N-terminal or inside the peptide chain.

2. Results and discussion

We began our program with the synthesis of [Sq-Tyr¹]-Leu-Enk **10a** (Scheme 1). To this end, the coupling precursor **14** was prepared using the Sq group-containing aminomalonate equivalent **11** (see Scheme 1).⁹ The alkylation reaction of **11** with bromide **12** in the presence of Bu₄NI and K₂CO₃ gave **13** in 74% yield. Due to the strong electron-withdrawing properties of the Sq group, it was expected that the carboxylic acid group in **13** would be removed by mild decarboxylation reaction conditions via the corresponding carboxylate.⁹ As expected, *t*-butyl ester **13** underwent the spontaneous decarboxylation reaction by treatment with TFA that produced **14**. The coupling reaction of **14** with H₂N-Gly-Gly-Phe-Leu-*O-t*-Bu was conducted in the presence of Et₃N followed by the removal of the protecting groups with HBr/AcOH to give Leu-enkephalin analog **10a** in high yield.

Next, we examined the synthesis of Leu-encephalin analogs **10b,c** possessing Sq-Gly inside the peptide chain. Our initial plan involved the use of Cbz-HN-Sq-Gly-*Oi*-Pr **15** as the coupling precursor, Eq. 1. However, the *N*-Cbz group could not be removed under the catalytic hydrogenation conditions since the Sq group significantly poisoned the catalyst activity, Eq. 1.¹² Attempts to synthesize other possible, such as *N*-Boc or *N*-Fmoc, Sq-Gly derivatives **17** could not be prepared by the

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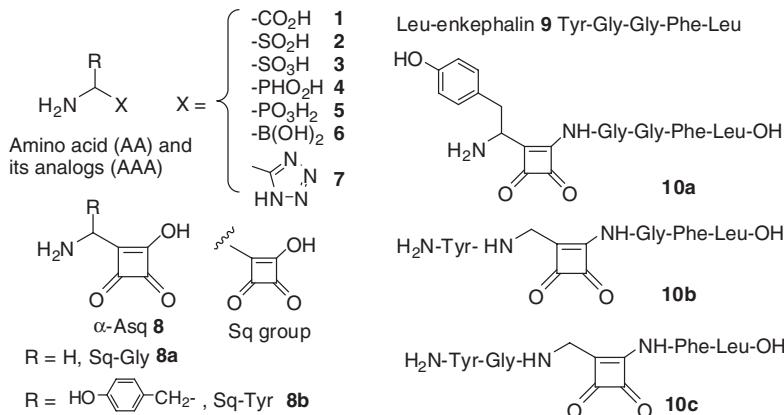
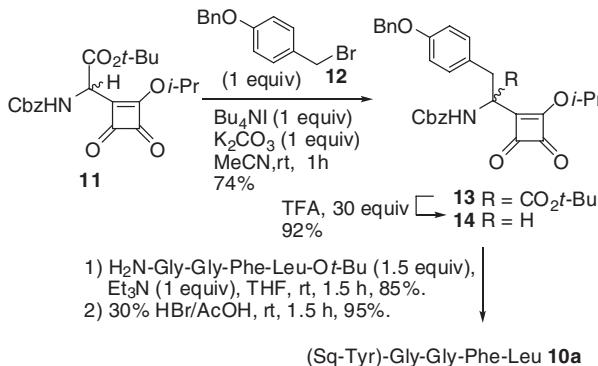
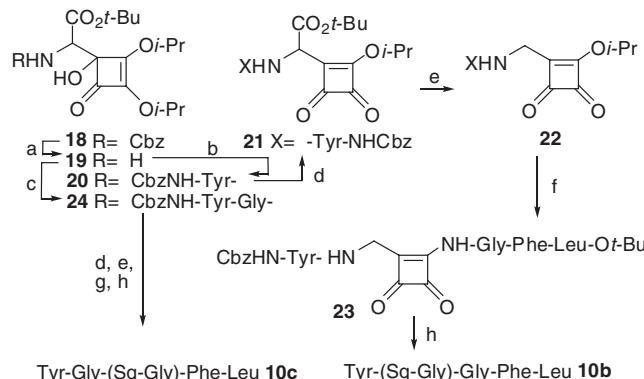
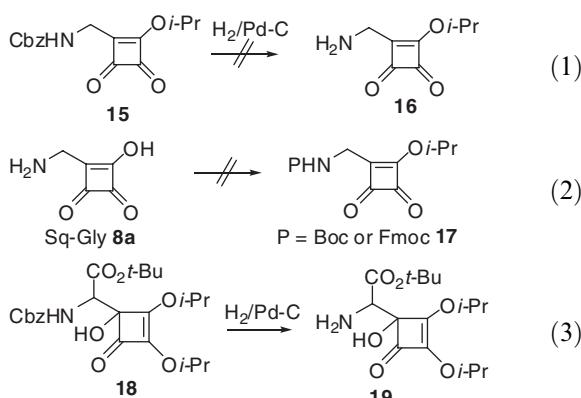


Figure 1. Structures of amino acid mimics Leu-Enkephalin analogs containing the α -Asqs.



Scheme 1. Synthesis of 10a.

introduction of protecting groups to Sq-Gly 8a or using the dianion enolate method, Eq. 2. Therefore, we turned our attention to hydroxybutenone 18^{9b,12} as an alternative coupling precursor, Eq. 3 since 18 underwent elimination of the Cbz group under $\text{H}_2/\text{Pd-C}$ conditions to generate the free amine 19, in contrast to the case of the Sq group-containing 16, Eq. 1.

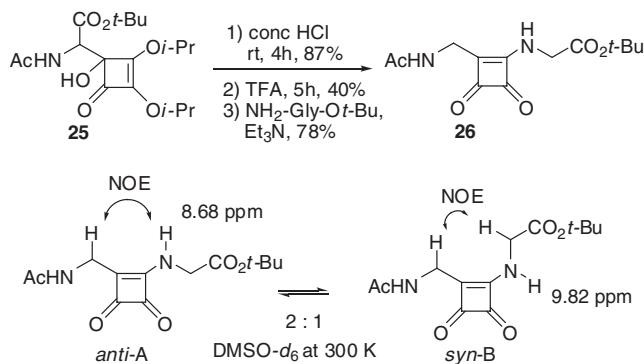


Scheme 2. Reagents and conditions: (a) Pd-C (20% w/w), H_2 , MeOH , rt, 2 h; (b) CbzNH-Tyr-OH (1.1 equiv), DEPBT , (2.2 equiv), THF , rt, 48 h, two steps, 78%; (c) CbzNH-Tyr-Gly-OH (1.1 equiv), DEPBT , (1.2 equiv), THF , rt, 48 h, two steps, 76%; (d) concd HCl , (0.9 equiv), CH_2Cl_2 , rt, 3 h, 74%; (e) TFA (30 equiv), CH_2Cl_2 , rt, 12 h, 74%; (f) $\text{H}_2\text{N-Gly-Phe-Leu-O-t-Bu}$ (1.5 equiv), Et_3N (1 equiv), THF , rt, 1.5 h, 91%; (g) $\text{H}_2\text{N-Phe-Leu-O-t-Bu}$ (1.5 equiv), Et_3N (1 equiv), THF , rt, 1.5 h, 91%; (h) 30% HBr/AcOH , rt, 1.5 h, 95–98%. Synthesis of 10b and 10c.

treatment with a small amount of concd HCl , (2) removal of the *t*-butyl ester by TFA, and (3) spontaneous decarboxylation of the corresponding carboxylate. Dipeptide 22 was coupled with $\text{H}_2\text{N-Gly-Phe-Leu-O-t-Bu}$ in the presence of Et_3N to give a protected pentapeptide analog 23. Removal of the protecting group of 23 with HBr/AcOH furnished [Sq-Gly²]-Leu-Enk 10b. In a similar manner, [Sq-Gly³]-Leu-Enk 10c was synthesized from 18 via 24 in good yield.

Along this line, 18 was converted to the corresponding amine 19, which, upon treatment with Cbz-Tyr-OH in the presence of DEPBT,¹³ afforded the coupling product 20 in 78% yield (Scheme 2). This was converted to dipeptide analog 22 via 21 by the following sequence of reactions: (1) conversion to cyclobutenedione 21 by

The binding affinities of 10a–c for the μ -, δ -, and κ -opioid receptors were investigated. But these analogs did not show any significant binding affinities.¹⁴ To gain insights into the structure and biological activity relationship of 10a–c, the conformational analysis of the α -Asq containing dipeptide analog 26 as a model of 10a–c was examined (see Scheme 3). Dipeptide 26, prepared from *N*-acetyl hydroxycyclobutenedione 25, provides certain information about the local conformation around the Asq moiety. The ¹H NMR data and NOE experiments of 26 in $\text{DMSO}-d_6$ indicated that 26 consisted of a 2:1



Scheme 3. Synthesis of **26** and its conformational analysis.

mixture of rotamers A (*anti*-A) and B (*syn*-B). The formation of a mixture of rotamers was also observed in **10a–c** and other Sq-amide derivatives as previously reported.¹⁵ The major rotamer was assigned to the *anti*-form A by the NOE experiments and by the comparison of the shift values of the amide protons of *syn*-B (9.82 ppm) and *anti*-A (8.68 ppm). The lower shift of the amide proton of *syn*-B was attributed to the paramagnetic influence of the carbonyl group in the Sq group. Measurement of the temperature coefficient values¹⁶ ($-d\delta/dT$, ppm/K) of each amide proton of **26** in DMSO-*d*₆ under variable temperatures showed large values ($-d\delta/dT > 0.0045$) for each amide proton. The results suggested that the solvent-shielded or intramolecular hydrogen bonding amide protons are not involved in this molecule. Enkephalin analogs **10a–c** were subjected to the same NMR studies. These results were quite similar to those of the model peptide **26** with respect to the formation of the rotamers in DMSO-*d*₆ and large temperature coefficient values of the amide protons ($-d\delta/dT > 0.0045$), indicating that the synthetic Asq analogs **10a–c** do not have a preferential binding conformation to the opioid receptors in which the β -turn conformation is proposed to be one of the crucial factors for the receptor binding.

In summary, we have established a synthetic route to access the α -Asq-containing enkephalins and evaluated their binding activities to opioid receptors. To the best of our knowledge, this is the first example incorporating α -Asq into peptide. Peptide analogs **10a–c** were found to be stable under acidic conditions, HPLC purification conditions, and NMR measurements in DMSO and D₂O. In contrast, peptide analogs embedding amino-phosphate **2** and aminosulfate **3** are known to be liable to hydrolytic cleavage.¹⁷ The chemically stable features of **10a–c** are advantageous for further application to the synthesis of other α -Asq-containing peptides.

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Supplementary data

Supplementary data associated with this article can be found, in the online version, at doi:10.1016/j.tetlet.2007.08.103.

References and notes

- (a) Dujols, F. J. M.; Mulliez, M. E. *J. Org. Chem.* **1996**, *61*, 5648–5649; (b) Merricks, D.; Sammes, P. G.; Walker, E. R. H.; Henrick, K.; McPartlin, M. M. *J. Chem. Soc., Perkin Trans. I* **1991**, 2169; (c) Gilmore, W. F.; Lin, H.-J. *J. Org. Chem.* **1978**, *43*, 4535.
- (a) Paik, S.; White, E. H. *Tetrahedron* **1996**, *52*, 5303; (b) Jinachitra, S.; MacLeod, A. J. *Tetrahedron* **1979**, *35*, 1315; (c) Shiba, T.; Miyoshi, K.; Kusumoto, S. *Bull. Chem. Soc. Jpn.* **1977**, *50*, 254; (d) Frankel, M.; Moses, P. *Tetrahedron* **1960**, *9*, 289; (e) Neelakantan, L.; Hartung, W. H. *J. Org. Chem.* **1959**, *24*, 1943.
- For recent examples: (a) Nasopoulou, M.; Matziari, M.; Dive, V.; Yiotakis, A. *J. Org. Chem.* **2006**, *71*, 9525; (b) Yang, K.-W.; Golich, F. C.; Sigdel, T. K.; Crowder, M. W. *Bioorg. Med. Chem. Lett.* **2005**, *15*, 5150; (c) Gurilingappa, H.; Buckhalts, P.; Kinzler, K. W.; Vogelstein, B.; Khan, S. R. *Bioorg. Med. Chem. Lett.* **2004**, *14*, 3531; (d) Mucha, A.; Paweleczak, M.; Hurek, J.; Kafarski, P. *Bioorg. Med. Chem. Lett.* **2004**, *14*, 3113; (e) Szabo, A.; Jaszay, Z. M.; Toke, L.; Petnehazy, I. *Tetrahedron Lett.* **2004**, *45*, 1991; (f) Matziari, M.; Beau, F.; Cuniasse, P.; Dive, V.; Yiotakis, A. *J. Med. Chem.* **2004**, *47*, 325; (g) Szabo, A.; Jaszay, Z. M.; Hegedus, L.; Toke, L.; Petnehazy, I. *Tetrahedron Lett.* **2003**, *44*, 4603; (h) Strancar, K.; Gobec, S. *Synthesis* **2004**, 359; (i) Kaboudin, B.; As-habe, N. *Tetrahedron Lett.* **2003**, *44*, 4243; (j) Grembecka, J.; Mucha, A.; Cierpicki, T.; Kafarski, P. *J. Med. Chem.* **2003**, *46*, 2641; (k) Rushing, S. D.; Hammer, R. P. *J. Am. Chem. Soc.* **2001**, *123*, 4861.
- For reviews: (a) Kafarski, P.; Lejczak, B. In *Aminophosphonic and Aminophosphinic Acids*; Kukhar, V. P., Hudson, H. R., Eds.; John Wiley and Sons: Chichester, 2000; (b) Kafarski, P.; Lejczak, B. *Curr. Med. Chem.—Anti-Cancer Agents* **2001**, *1*, 301.
- For recent synthetic examples of α -amino phosphonic acids: (a) Li, B.; Cai, S.; Du, D.; Xu, J. *Org. Lett.* **2007**, *9*, 2257; (b) Fernandez, M. C.; Diaz, A.; Guillen, J. J.; Blanco, O.; Ruiz, M.; Ojea, V. *J. Org. Chem.* **2006**, *71*, 6958; (c) Xu, J.; Gao, Y. *Synthesis* **2006**, 783; (d) Liu, H.; Cai, S.; Xu, J. *J. Pept. Sci.* **2006**, *12*, 337; (e) Swamy, K. C. K.; Kumaraswamy, S.; Kumar, K. S.; Muthiah, C. *Tetrahedron Lett.* **2005**, *46*, 3347; (f) Kobayahi, S.; Kiyohara, H.; Nakamura, Y.; Matsubara, R. *J. Am. Chem. Soc.* **2004**, *126*, 6558; (g) Joly, G. D.; Jacobsen, E. N. *J. Am. Chem. Soc.* **2004**, *126*, 4102; (h) Kaboudin, B.; Saadati, F. *Synthesis* **2004**, 1249; (i) Tongcharoensirikul, P.; Suarez, A. I.; Voelker, T.; Thompson, C. M. *J. Org. Chem.* **2004**, *69*, 2322; (j) de Medina, P.; Ingrassia, L. S.; Mulliez, M. E. *J. Org. Chem.* **2003**, *68*, 8424; (k) Davis, F. A.; Prasad, K. R. *J. Org. Chem.* **2003**, *68*, 7249; (l) Chen, X.; Wiemer, D. F. *J. Org. Chem.* **2003**, *68*, 6108; (m) Azizi, N.; Saidi, M. *Tetrahedron* **2003**, *59*, 5329; (n) Palacios, F.; Aparicio, D.; Ochoa de Retana, A. M.; de los Santos, J. M.; Gil, J. I.; Lopez de Munain, R. *Tetrahedron: Asymmetry* **2003**, *14*, 689; (o) Chandrasekhar, S.; Narasimulu,

- C. H.; Sultana, S. S.; Saritha, B.; Prakash, S. J. *Synlett* **2003**, 505.
6. For a review: Dembitsky, V. M.; Srebnik, M. *Tetrahedron* **2003**, 59, 579.
7. (a) Herr, R. J. *Bioorg. Med. Chem.* **2002**, 10, 3379; (b) Demko, Z. P.; Sharpless, K. B. *Org. Lett.* **2002**, 4, 2525; (c) Boteju, L. W.; Hruby, V. J. *Tetrahedron Lett.* **1993**, 34, 1757; (d) Zabrocki, J.; Dunbar, J. B., Jr.; Marshall, K. W.; Toth, M. V.; Marshall, G. R. *J. Org. Chem.* **1992**, 57, 202.
8. For selected examples: (a) Hischmann, R.; Smith, A. B., III; Taylor, C. M.; Benkovic, P. A.; Taylor, S. D.; Yager, K. M.; Sprengeler, P. A.; Benkovic, S. J. *Science* **1994**, 265, 234; (b) Schultz, P. G. *Angew. Chem., Int. Ed. Engl.* **1989**, 10, 1283; (c) Schultz, P. G. *Acc. Chem. Res.* **1989**, 22, 287; (d) Bartlett, P. A.; Acher, F. *Bull. Chim. Soc., Fr.* **1986**, 771.
9. (a) Shinada, T.; Ishida, T.; Ohfune, Y. *Synthesis* **2005**, 2723; (b) Shinada, T.; Ishida, T.; Ohfune, Y. *Tetrahedron Lett.* **2005**, 45, 311.
10. For reviews: (a) West, R.; Niu, J. In *Nonbenzenoid Aromatics*; Snyder, J. P., Ed.; Academic Press: New York, 1969; p 311; (b) West, R. *Oxocarbons*; Academic Press: New York, 1980; (c) Schmidt, A. H. *Synthesis* **1980**, 961; (d) Seitz, G.; Imming, P. *Chem. Rev.* **1992**, 2, 1227; (e) Ohno, M.; Eguchi, S. In *Bioactive Heterocycles I*; Eguchi, S., Ed.; Topics in Heterocyclic Chemistry; Springer: Berlin, 2006; Vol. 6, pp 1–37.
11. Recent examples of squaric acid derivatives in bioorganic and medicinal chemistries: (a) Butera, J. A.; Jenkins, D. J.; Lennox, J. R.; Sheldon, J. H.; Norton, N. W.; Warga, D.; Argentieri, T. M. *Bioorg. Med. Chem. Lett.* **2005**, 15, 2495; (b) Lee, C.-W.; Cao, H.; Ichiyama, K.; Rana, T. M. *Bioorg. Med. Chem. Lett.* **2005**, 15, 4243; (c) Onaran, M. B.; Comeau, A. B.; Seto, C. T. *J. Org. Chem.* **2005**, 70, 10792; (d) Smet, C.; Duckert, J.-F.; Wieruszkeski, J.-M.; Landrieu, I.; Buee, L.; Lippens, G.; Deprez, B. *J. Med. Chem.* **2005**, 48, 4815; (e) Xie, J.; Comeau, A. B.; Seto, C. T. *Org. Lett.* **2004**, 6, 83; (f) Valgeirsson, J.; Nielsen, E. O.; Peters, D.; Mathiesen, C.; Kristensen, A. S.; Madsen, U. *J. Med. Chem.* **2004**, 47, 6948; (g) Sun, L.; Chiu, D.; Kowal, D.; Simon, R.; Smeyne, M.; Zukin, R. S.; Oley, J.; Baudy, R.; Lin, S. *J. Pharm. Exp. Ther.* **2004**, 310, 563; (h) Saksena, R.; Chernyak, A.; Poirot, E.; Kovac, P. *Methods Enzymol.* **2003**, 362, 140; (i) Shinada, T.; Nakagawa, Y.; Hayashi, K.; Corzo, G.; Nakajima, T.; Ohfune, Y. *Amino Acids* **2003**, 24, 293; (j) Lim, N. C.; Morton, M. D.; Jenkins, H. A.; Bruckner, C. *J. Org. Chem.* **2003**, 68, 9233; (k) Sato, K.; Seio, K.; Sekine, M. *J. Am. Chem. Soc.* **2002**, 124, 12715; (l) Porter, J. R.; Archibald, S. C.; Childs, K.; Critchley, D.; Head, J. C.; Linsley, J. M.; Parton, T. A. H.; Robinson, M. K.; Shock, A.; Taylor, R. J.; Warrelow, G. J.; Alexander, R. P.; Langham, B. *Bioorg. Med. Chem. Lett.* **2002**, 12, 1051.
12. Shinada, T.; Hayashi, K.; Yoshida, Y.; Ohfune, Y. *Synlett* **2000**, 1506.
13. Li, H.; Xiaohui, J.; Yun-hua, Y.; Chongxu, F.; Romoff, T.; Goodman, M. *Org. Lett.* **1999**, 1, 91.
14. Minami, M.; Satoh, M. *Neurosci. Res.* **1995**, 23, 121. IC₅₀ values were determined by the competitive inhibition of the specific binding of 1 nM [³H] [D-Ala, N-Me-Phe⁴, Gly-ol⁵]-enkephalin, 1 nM [³H] [D-penilliniamine^{2,5}]-enkephalin, or 2 nM [³H] (**5a**, **7a**, **8b**)-(+)N-methyl-N-[7-(1-pyrrolidinyl)-1-oxaspiro[4,5]dec-8-yl-benzeneacetamide in a membrane prepared from CHO cells expressing the μ -, δ -, or κ -receptor, respectively. Binding affinities of Leu-Enk and **10a–c** [Leu-Enk: μ -receptor (1.8 nM), δ -receptor (0.6 nM), κ -receptors (400 nM); **10a**: μ -receptor (>1000 nM), δ -receptor (>1000 nM), κ -receptors (>1000 nM); **10b**: μ -receptor (>1000 nM), δ -receptor (420 nM), κ -receptors (>1000 nM); **10c**: μ -receptor (350 nM), δ -receptor (970 nM), κ -receptors (>1000 nM)].
15. Rotger, M. C.; Pina, M. N.; Frontera, A.; Martorell, G.; Ballester, P.; Deya, P. M.; Costa, A. *J. Org. Chem.* **2004**, 69, 2302.
16. (a) Horikawa, M.; Shigeri, Y.; Yumoto, N.; Yoshikawa, S.; Nakajima, T.; Ohfune, Y. *Bioorg. Med. Chem. Lett.* **1998**, 8, 2027; (b) Kishore, R.; Balaram, P. *Indian J. Chem.* **1984**, B, 1137; (c) Ravi, A.; Venkataram Prasad, B. V.; Balaram, P. *J. Am. Chem. Soc.* **1983**, 105, 105; (d) Iqbal, M.; Balaram, P. *J. Am. Chem. Soc.* **1981**, 103, 5548.
17. The stability depended on the pH conditions: (a) Mucha, A.; Kunert, A.; Grembecka, J.; Pawełczak, M.; Kafarski, P. *Eur. J. Med. Chem.* **2006**, 41, 768; (b) Yiotakis, A.; Lecocq, A.; Nicolaou, A.; Labadie, J.; Dive, V. *Biochem. J.* **1994**, 303, 327; (c) Hanson, J. E.; Kaplan, A. P.; Bartlett, P. A. *Biochemistry* **1989**, 28, 6294; (d) Yamauchi, K.; Ohtsuki, S.; Kinoshita, M. *J. Org. Chem.* **1984**, 49, 1158.